STUDIES ON NEW DEHYDROPEPTIDASE INHIBITORS

I. TAXONOMY, FERMENTATION, ISOLATION AND PHYSICO-CHEMICAL PROPERTIES

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WS1358A1 (FR104007) and B1 (FR104008), new potent inhibitors of renal dehydropeptidase, were isolated from the culture broth of strain No. 1358 which was identified as *Streptomyces parvulus* subsp. *In vitro* inhibitory activities (IC₅₀ value) of WS1358A1 and B1 against porcine renal DHP were 3 and 600 nm, respectively.

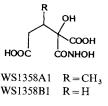
Renal dipeptidase, dehydropeptidase^{1,2)} (DHP, EC.3.4.13.11.) hydrolyzes the β -lactam ring of the carbapenems and penems which are new classes of structurally novel β -lactam antibiotics^{3~5)}. The low recovery of the carbapenems in the urine of laboratory animal and the low efficacy against experimental infection are attributed to the metabolism of the antibiotics by DHP⁶⁾.

Cilastatin^{7,8)}, an inhibitor of DHP, was synthesized by Merck's researchers and has been clinically used in a new combination antimicrobial imipenem/cilastatin. In this combination, cilastatin has been proved to be effective in protecting imipenem, a derivative of thienamycin, from hydrolysis by DHP and restoring the urinary recovery of imipenem in human as well as in experimental animals⁹⁾.

No specific inhibitor against DHP other than cilastatin has been yet reported. During our screening program searching for novel enzyme inhibitors, we recently found specific inhibitors against DHP, designated WS1358A1 and B1^{\dagger} (Fig. 1), from the fermentation products.

This paper describes taxonomic studies on the producing strain, fermentation, isolation and

Fig. 1. Structures of WS1358A1 and B1.



physico-chemical properties of these compounds. The structural elucidation of the compounds will be described in a succeeding $paper^{10}$.

Materials and Methods

Taxonomic Studies

The producing organism, strain No. 1358, was isolated from a soil sample collected at Tochigi city, Tochigi Prefecture, Japan. The media and procedures used for cultural and physiological characterization of strain No. 1358 were described by SHIRLING and GOTTLIEB¹¹. Each culture was incubated at 30°C for

[†] WS1358A1 and B1 have been reported as FR104007 and FR104008, respectively, in Eur. Pat. Appl. 276, 947, Aug. 3, 1989.

2 to 3 weeks before observation. The temperature range for growth was determined on yeast extract - malt extract agar using a temperature gradient incubator (Advantec Toyo Co.). The color names used in these studies were based on the Methuen Handbook of Colour^{12}). The chemical composition of the cell wall was analyzed by the methods described by BECKER *et al.*¹³⁾, and LECHEVALIER and LECHEVALIER¹⁴⁾. Utilization of carbon sources was examined according to the method of PRIDHAM and GOTTLIEB¹⁵⁾.

Fermentation

A loopful of the slant culture of strain No. 1358 was inoculated to a 500-ml flask containing 160 ml of the seed medium composed of corn starch 1.0%, glycerol 1.0%, glucose 0.5%, cotton-seed flour 1.0%, dried yeast 0.5%, corn steep liquor 0.5% and $CaCO_3$ 0.2% (pH 6.5). The flask were shaken on a rotary shaker (220 rpm) for 3 days at 30°C. A 200-liter jar fermenter containing 150 liters of production medium composed of glucose 1.0%, glycerol 2.0%, cotton-seed flour 1.0%, soy bean meal 1.0%, dried yeast 0.5% and $CaCO_3$ (pH 7.0) was inoculated with 3 liters of the seed broth and cultured for 4 days at 30°C, aerated at 150 liters per minute and agitated at 250 rpm.

Enzyme Assay

The production and initial purification of WS1358 compounds were monitored with the enzyme assay to determine the inhibitory activity. Partially purified porcine renal DHP is able to hydrolyze the unsaturated dipeptide, glycyldehydrophenylalanine (GDP) as substrate¹⁶⁾. The activity of the enzyme was determined by observing the decline in absorbance of the substrate at 275 nm. Inhibitory effect of a compound on DHP activity was also measured in a similar way. Inhibition percent, I%, was calculated as follows, $I\% = (E-T)/E \times 100$, where E was DHP activity without the compound, T was DHP activity with the compound.

Materials

Partial purification of DHP from porcine kidney cortex was carried out using procedures slightly different from a previously reported method²⁾. Details will be described in a separate paper. GDP was synthesized by the method previously described¹⁶⁾. Cilastatin was purified from a commercially available drug, Zienam.

Results

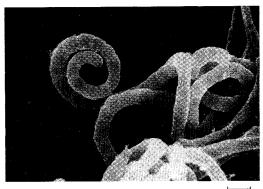
Identification of Strain No. 1358

Morphological observations were made with a light and a scanning electron microscope (Fig. 2) on cultures grown at 30°C for 2 to 3 weeks on yeast extract-malt extract agar, oatmeal agar and inorganic

salts - starch agar¹¹⁾. The substrate mycelia were well developed and branched without fragmentation. The aerial mycelia branched monopodially, and formed spiral, sometimes looped, spore chains with 10 to 50 spore per chain. The spores had a smooth surface and were subglobose to oblong in shape with a size of 0.5 to 0.7 by 0.5 to $0.9 \,\mu$ m. These spore masses often appeared to coalesce in moist globules. Sclerotic granules, sporangia and zoo-spores were not observed. The cultural characteristics of strain No. 1358 are summarized in Table 1. The aerial mass color was brownish gray to grayish brown. Reverse side of growth was yellowish white to

Fig. 2. Scanning electron microphotography of aerial mycelia of strain No. 1358.

The bar represents $1 \,\mu m$.



Medium	Growth	Aerial mycelium	Reverse side of substrate mycelium	Soluble pigments
Yeast extract - malt extract agar	Abundant	White, brownish gray (6D2), grayish brown (7F3)	Grayish yellow (4B3)	None
Oatmeal agar	Moderate	Grayish brown (5F3)	Yellowish white (3A2)	None
Inorganic salts - starch agar	Abundant	Brownish gray (7D-E2)	Grayish yellow (4B3)	None
Glycerol - asparagine agar	Abundant	Brownish gray (6C2)	Yellowish white (4A2)	None
Peptone - yeast extract - iron agar	Moderate	None	Light yellow (4A4-5)	None
Tyrosine agar	Abundant	Orange gray (5B2), brownish gray (7E2)	Grayish orange (5B3)	Scant, grayish orange
Glucose - asparagine agar	Abundant	Brownish gray (6C-D2), grayish brown (7F3)	Yellowish white (4A2)	None
Nutrient agar	Moderate	None	Pale yellow (3A3)	None
BENNETT agar	Abundant	Dark brown (6F4)	Pale yellow (3A3)	None
Sucrose - nitrate agar	Abundant	None	Pale yellow (3A3)	None

Table 1. Cultural characteristics of strain No. 1358.

Table 2. Physiological characteristics of strain No. 1358.

Temperature range for growth	16~33°C
Optimum temperature for growth	26∼32°C
Liquefaction of gelatin	Weakly positive
Coagulation of milk	Negative
Peptonization of milk	Weakly positive
Hydrolysis of starch	Positive
Melanoid production	Negative
Decomposition of cellulose	Negative
Nitrate reduction	Positive
NaCl tolerance	$0 \sim 2\%$

grayish yellow. Melanoid and other soluble pigments were not produced. The physiological characteristics and utilization of carbon sources of strain No. 1358

Table 3. Utilization of carbon sources by strain No. 1358.

Carbon source	Utilization
Glucose	+
L-Arabinose	+
D-Xylose	+
Inositol	+
Mannitol	+
D-Fructose	+
L-Rhamnose	+
Sucrose	+
Raffinose	+
No addition	_

-: No growth, +: good growth.

are summarized in Tables 2 and 3. Hydrolyzed whole cell of strain No. 1358 contained LL-diaminopimelic acid. Accordingly, the cell wall of this strain is believed to be of type I.

Based on the taxonomic properties described above, strain No. 1358 is considered to belong to the genus *Streptomyces*¹⁷⁾ and to be a strain of the gray series of the PRIDHAM and TRESNER grouping¹⁸⁾. Strain No. 1358 was compared with *Streptomyces* species described in the literature^{19~23)}. As a result, it was found that the strain proved to closely resemble *Streptomyces parvulus* IFO 13193 in detail. There, it was found that the properties of both strains were almost identical except for a few differences. Table 4 shows the differences between the two strains. These differences are not sufficient to consider that strain No. 1358 belong to a distinct species. So, it is considered to be proper that strain No. 1358 is a sub-species strain of *S. parvulus*. Therefore, this strain was designated as *S. parvulus* subsp. *tochigiensis* No. 1358. The strain has been deposited at the Fermentation Research Institute, Agency of Industrial Science and Technology, Japan under the accession No. FERM BP-1638.

	No. 1358	IFO 13193
Temperature range for growth (°C)	16~33	14~40
NaCl tolerance (%)	2	10

Table 4. Differences between strain No. 1358 and Streptomyces parvulus IFO 13193.

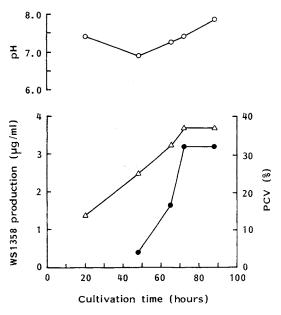
Production of WS1358 Compounds

A typical fermentation profile for the production of WS1358 is shown in Fig. 3. The production of active compounds during the fermentation was monitored by measuring the inhibitory activity against DHP and assessed from the standard curve of the purified WS1358A1. The production as well as the cell growth was initiated on day 2 and rapidly reached a maximum of about $3.2 \,\mu$ g/ml on day 3.

Isolation and Purification of WS1358A1 and B1

The procedures of the isolation are summarized in Fig. 4. Throughout these purification procedures, the WS1358 compounds were monitored by an analytical reverse-phase HPLC instead of enzyme assay. Fig. 3. A typical time course of fermentation by strain No. 1358.

• Potency, \bigcirc pH, \triangle packed cell volume (PCV).



The potency indicates the total activities of WS1358 compounds in broth filtrate as that of WS1358A1.

It is very difficult to isolate WS1358A1 directly from the fermentation broth by means of conventional methods, because it is only produced in a small quantity and has the some physico-chemical properties close to those of some impurities. However, we discovered that WS1358A1 can be reversibly transformed to the two compounds, designated A2 and A3, which are more lipophilic and more easily isolated from the fermentation broth than A1 is. Making use of this chemical transformation to purify WS1358A1, we converted A1 to A2 and A3 in the course of purification and finally converted the purified A3 to A1. More detailed mechanism of this chemical transformation will be described elsewhere.

The cultured broth was filtered with the aid of diatomaceous earth. The filtrate (390 liters) was adjusted to pH 10 with $6 \times NaOH$ and allowed to stand for 2 hours at room temperature (A2, A3 \rightarrow A1 conversion). The resultant precipitate was filtered and discarded. The filtrate was passed through a column of Dowex 1X2 (Cl⁻ type). The active principle was eluted with 0.2 M NaCl solution after washing the column with water and 0.1 M NaCl solution. The eluate was adjusted to pH 2 with $6 \times HCl$ and adsorbed on activated carbon. The active principle was eluted with 25% aqueous methanol containing 0.5 N NH₄OH after washing the column with water and 25% aqueous methanol. The eluate was concentrated *in vacuo* and adjusted to pH 2 with $6 \times HCl$. The desalted eluate was applied on a column of an adsorption resin Diaion SP-207 and developed with water. The fraction containing WS1358A1 was eluted more slowly than the fraction containing WS1358B1.

WS1358A3

The former fraction of the Diaion SP-207 column chromatography was concentrated in vacuo followed

Fig. 4. Isolation procedure of WS1358 compounds.

Culture filtrate (390 liters)

adjusted to pH 10, filtered

Filtrate

Dowex 1X2 (Cl⁻)

eluted with 0.2 M NaCl adjusted to pH 2

Active carbon

eluted with 25 % aq MeOH – 0.5 \varkappa NH4OH concentrated

adjusted to pH 2

Diaion SP-207

eluted with H₂O

neutralized concentrated

Cellulose

eluted with 75 % aq 2-PrOH concentrated

DEAE Sephadex A-25 (Cl⁻)

eluted with 0.2 M NaCl concentrated

Sephadex G-15

eluted with H₂O concentrated

B1 (1.3g)

neutralized concentrated added with an equal volume of 0.2 M NH4H2PO4 (pH 2.3) Dowex 1X2 (Cl⁻)

eluted with 0.1 M NH4H2PO4 freeze-dried

Diaion SP-207

eluted with H₂O

concentrated

Sephadex G-15

eluted with H₂O freeze-dried

A3 (30.2 mg)

dissolved in 0.5 N NaOH neutralized

DEAE Sephadex A-25 (Cl-)

eluted with 0.2 M NaCl concentrated adjusted to pH 2

Diaion SP-207

eluted with H₂O neutralized freeze-dried

A1 (14.9 mg)

Sephadex G-15

eluted with H₂O freeze-dried

A2 (28 mg)

by the addition of an equal volume of $0.2 \text{ M } \text{NH}_4\text{H}_2\text{PO}_4$ buffer (pH 2.3). The resultant solution was passed through a column of Dowex 1X2 (Cl⁻ type) equilibrated with $0.1 \text{ M } \text{NH}_4\text{H}_2\text{PO}_4$ buffer and eluted with the same buffer. The active fraction containing WS1358A1 was concentrated *in vacuo* and freeze-dried (A1 \rightarrow A2+A3 conversion). The lyophilized material thus obtained was dissolved in deionized water. The solution was applied on a column of Diaion SP-207 and developed with water. WS1358A2 and A3 could be separated from each other in this column chromatography. The fraction containing WS1358A3 was neutralized with 6 N NaOH and applied on a column of Sephadex G-15. The eluate was freeze-dried to give a white powder of WS1358A3 (35.4 mg).

WS1358A1

WS1358A3 (30.2 mg) was dissolved in 0.5 N NaOH (3.5 ml) and allowed to stand for 30 minutes at room temperature (A3 \rightarrow A1 conversion), then diluted with water after neutralization with 6 N HCl. The

	A1	B1
MP (°C, dec)	98~100	92~93
Molecular formula	$C_7H_9NO_7Na_2$	$C_6H_7NO_7Na_2$
Elementary analysis	Calcd for $C_7H_9NO_7Na_2 \cdot H_2O$	Calcd for $C_6H_7NO_7Na_2 \cdot H_2O$
Caled:	C 29.69, H 3.92, N 4.95, Na 16.24	C 26.78, H 3.37, N 5.20, Na 17.08
Found:	C 30.05, H 3.84, N 4.88, Na 15.89	C 26.44, H 3.28, N 5.10, Na 16.88
FAB-MS (m/z) for free acid	222 $(M^+ + 1)$	$208 (M^+ + 1)$
$[\alpha]_{D}^{23}$ (H ₂ O)	$-14.0^{\circ} (c \ 0.9)$	$+2.5^{\circ}$ (c 1.0)
UV $\lambda_{\max}^{H_2O}$ nm (ε)	End absorption	End absorption
IR v_{max} (KBr) cm ⁻¹	3500~2500, 1660, 1580, 1360, 1160,	3500~2500, 1660, 1620, 1580, 1400,
	1100, 1000, 880, 800	1380, 1200, 1130, 1100, 880, 800
Rf value ^a (I)	0.43	0.40
(II)	0.56	0.46

Table 5. Physico-chemical properties of WS1358A1 and B1.

^a Silica gel TLC (Merck Art. No. 5715), solvent (I) BuOH - AcOH - H₂O (2:1:1), (II) 2-PrOH - H₂O (65:35).

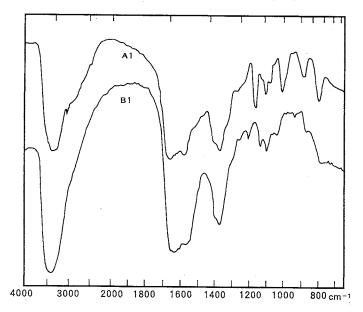


Fig. :	5.	IR spectra	of WS1358A1	and B1	(KBr).
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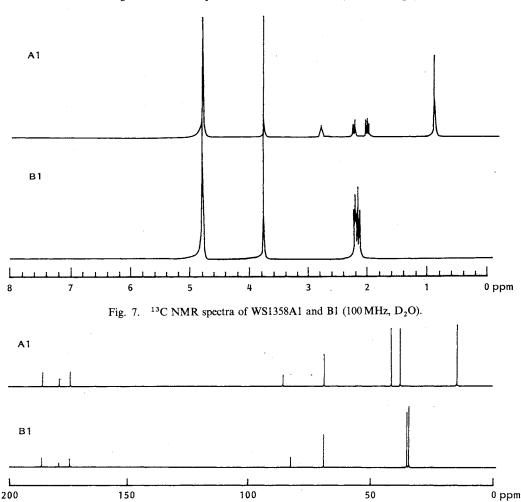
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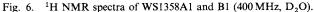
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resultant solution was applied on a column of DEAE-Sephadex A-25 (Cl⁻ type) and eluted with water, 0.1 M NaCl and 0.2 M NaCl solution. The active fraction was adjusted to pH 2 with 6 N HCl and concentrated *in vacuo*, then applied on a column of Diaion SP-207 and developed with water. The eluate was neutralized with 1 N NaOH and freeze-dried to give a white powder of WS1358A1 as Na salt (14.9 mg).

WS1358B1

The fraction containing WS1358B1 obtained by the 1st Diaion SP-207 column chromatography was neutralized with 1×100 and concentrated *in vacuo*. The resultant solution was applied on a column of Cellulose CF11 and washed with 2-propanol, then developed with 75% aqueous 2-propanol. The active fraction was concentrated *in vacuo* and applied on a column of DEAE-Sephadex A-25 (Cl⁻ type). The active fraction was eluted with 0.2×1000 NaCl solution and concentrated *in vacuo*. The eluate was applied on a column of Sephadex G-15 equilibrated with water, then eluted with water. The desalted eluate was concentrated *in vacuo* to give the residue. The residue was recrystallized from aqueous methanol to give a white powder of WS1358B1 as Na salt (1.3 g).





Physico-chemical Properties of WS1358A1 and B1

Physico-chemical properties and spectral data of WS1358A1 and B1 Na salts are summarized in Table 5. The two compounds are soluble in water, slightly soluble in methanol and insoluble in acetone, ethyl acetate, chloroform and *n*-hexane. They give positive color reactions for ferric chloride, Ehrlich, ninhydrin and iodine vapor reagents, but not for Dragendorff, Molisch and cerium sulfate reagents. These data suggested that the compounds have *N*-hydroxyl group. The IR, ¹H and ¹³C NMR spectra of Na salts are shown in Figs. 5, 6 and 7, respectively. The structures of WS1358A1 and B1 were deduced as shown in Fig. 1 on the basis of the physico-chemical and spectral data. Details of the structural elucidation will be described in the following paper¹⁰.

Biological Properties of WS1358A1 and B1

WS1358A1 and B1 were found to strongly inhibit DHP from porcine kidney cortex. The IC₅₀ values of WS1358A1 and B1 were estimated to be about 3 and 600 nm respectively while that of cilastatin was 130 nm. They exhibited no antimicrobial activity at the concentration of 1 mg/ml by the disk diffusion assay method against *Escherichia coli*, *Staphylococcus aureus*, *Bacillus subtilis*, *Candida albicans* and *Aspergillus niger*. The acute toxicity (LD₅₀) of WS1358A1 was >1 g/kg when administrated intravenously to mice.

Discussion

The susceptibility of carbapenem antibiotics to metabolic inactivation by renal DHP is a major hindrance to the development of these antibiotics²⁴). There are two ways to overcome this problem. One way is to combine the carbapenem with a DHP inhibitor, the other way is to synthesize derivatives²⁵) of carbapenem which are resistant to enzymatic inactivation by DHP. Merck's researchers resolved the problem by synthesizing an DHP inhibitor, cilastatin⁷). The antimicrobial combination imipenem/cilastatin has been proved to be very effective⁹.

From the same point of view, we have been searching for novel DHP inhibitors of microbial origin in place of cilastatin. Recently we discovered potent DHP inhibitors, WS1358A1 and B1, produced by *Streptomyces* sp. They showed no cytotoxicity to the cultured cells (data not shown) and no toxicity to experimental animals.

In particular, WS1358A1 has more potent inhibitory activity than cilastatin and may have the potential to be developed in an antimicrobial combination with a carbapenem or penem antibiotic.

References

- GREENSTEIN, J. P.: Advances in Enzymology and Related Subjects of Biochemistry. Ed., F. F. NORD, pp. 117~169, Interscience Publish, Inc., 1948
- CAMPBELL, B. J.: Renal dipeptidase. In Methods in Enzymology. Vol. 19. Ed., S. P. COLOWICK et al., pp. 722~729, Academic Press, 1970
- CAMPBELL, B. J.; L. J. FORRESTER, W. L. ZAHLER & M. BURKS: β-Lactamase activity of purified and partially characterized human renal peptidase. J. Biol. Chem. 259: 14586~14590, 1984
- KAHAN, F. M.; H. KROPP, J. G. SUNDELOF & J. BIRNBAUM: Thienamycin: development of imipenem-cilastatin.
 J. Antimicrob. Chemother. 12 (Suppl. D): 1~35, 1983
- 5) CASSINELLI, G.; R. CORIGLI, P. OREZZI, G. VENTRELLA, A. BEDESCHI, E. PERRONE, D. BORGHI & G. FRANCESCHI: Structure determination of the primary renal metabolite of the penem FCE 22101. J. Antibiotics 41: 984 ~ 987, 1988
- 6) KROPP, H.; J. G. SUNDELOF, R. HAJDU & F. M. KAHAN: Metabolism of thienamycin and related carbapenem antibiotics by the renal dipeptidase, dehydropeptidase-I. Antimicrob. Agents Chemother. 22: 62~70, 1982
- 7) ASHTON, W. T.; L. BARASH, J. E. BROWN, R. D. BROWN, L. F. CANNING, A. CHEN, D. W. GRAHAM, F. M. KAHAN, H. KROPP, J. G. SUNDELOF & E. F. ROGERS: Z-2-Acylamino-3-substituted propenoates, inhibitors of the renal dipeptidase (dehydropeptidase-I) responsible for thienamycin metabolism. Program and Abstracts of the 20th

Intersci. Conf. on Antimicrob. Agents Chemother., No. 271, New Orleans, Sept. 22~24, 1980

- GRAHAM, D. W.; W. T. ASHTON, L. BARASH, J. E. BROWN, R. D. BROWN, L. F. CANNING, A. CHEN, J. P. SPRINGER & E. F. ROGERS: Inhibition of the mammalian β-lactamase renal dipeptidase (dehydropeptidase-I) by (Z)-2-(acylamino)-3-substituted-propenoic acids. J. Med. Chem. 30: 1074~1090, 1987
- 9) NORRBY, S. R.; K. ALESTIG, B. BJÖRNEGARD, L. Å. BURMAN, F. FERBER, J. L. HUBER, K. H. JONES, F. M. KAHAN, J. S. KAHAN, H. KROPP, M. A. P. MEISINGER & J. G. SUNDELOF: Urinary recovery of N-formimidoyl thienamycin (MK0787) as affected by coadministration of N-formimidoyl thienamycin dehydropeptidase inhibitors. Antimicrob. Agents Chemother. 23: 300~307, 1983
- TAKASE, S.; I. UCHIDA, S. HASHIMOTO, H. TANAKA & M. HASHIMOTO: Studies on new dehydropeptidase inhibitors. II. Structural elucidation and synthesis of WS1358A1 and B1. J. Antibiotics 43: 38~42, 1990
- 11) SHIRLING, E. B. & D. GOTTLIEB: Methods for characterization of *Streptomyces* species. Int. J. Syst. Bacteriol. 16: 313~340, 1966
- 12) KORNERUP, A. & J. H. WANSCHER (Ed.): Methuen Handbook of Colour. Eyre Methuen Ltd., 1978
- BECKER, B.; M. P. LECHEVALIER & H. A. LECHEVALIER: Chemical composition of cell-wall preparations from strains of various form-genera of aerobic actinomycetes. Appl. Microbiol. 13: 236~243, 1965
- LECHEVALIER, M. P. & H. A. LECHEVALIER: The chemotaxonomy of actinomycetes. In Actinomycete Taxonomy. Eds., A. DIETZ & D. W. THAYER, pp. 227~291, Society for Industrial Microbiology, 1980
- PRIDHAM, T. G. & D. GOTTLIEB: The utilization of carbon compounds by some Actinomycetales as an aid for species determination. J. Bacteriol. 56: 107~114, 1948
- 16) CAMPBELL, B. J.; Y. C. LIN & M. E. BIRD: Reactions of renal dipeptidase and cupric ions with unsaturated dipeptides J. Biol. Chem. 238: 3632~3639, 1963
- 17) WAKSMAN, S. A. (Ed.): The Actinomycetes. Vol. 2. Classification, Identification and Description of Genera and Species. Williams & Wilkins Co., 1961
- 18) PRIDHAM, T. G. & H. D. TRESNER: Genus I. Streptomyces Waksman and Henrici, 1943, 339. In BERGEY'S Manual of Determinative Bacteriology. 8th Ed. Eds., R. E. BUCHANAN & N. E. GIBBONS, pp. 748 ~ 829, Williams & Wilkins Co., 1974
- SHIRLING, E. B. & D. GOTTLIEB: Cooperative description of type culture of *Streptomyces*. II. Species descriptions from first study. Int. J. Syst. Bacteriol. 18: 69~189, 1968
- 20) SHIRLING, E. B. & D. GOTTLIEB: Cooperative description of type culture of *Streptomyces*. III. Additional species descriptions from first and second studies. Int. J. Syst. Bacteriol. 18: 279~392, 1968
- SHIRLING, E. B. & D. GOTTLIEB: Cooperative description of type culture of *Streptomyces*. IV. Species descriptions from the second, third and fourth studies. Int. J. Syst. Bacteriol. 19: 391 ~ 512, 1969
- 22) SKERMAN, V. B. D.; V. MCGOWAN & P. H. A. SNEATH: Approved lists of bacterial names. Int. J. Syst. Bacteriol. 30: 225~420, 1980.
- 23) MOORE, W. E. C.; E. P. CATO & L. V. H. MOORE: Index of the bacterial and yeast nomenclatural changes published in *International Journal of Systematic Bacteriology* since the 1980 approved list of bacterial names. (1 January 1980 to 1 January 1985). Int. J. Syst. Bacteriol. 35: 382~407, 1985
- 24) FOLLATH, F.; A. M. GEDDES, P. SPRING, G. D. BALL, K. H. JONES, F. FERBER, J. S. KAHAN & F. M. KAHAN: Tolerability and pharmacokinetics of single doses of N-formimidoyl-thienamycin. Program and Abstracts of the 21st Intersci. Conf. on Antimicrob. Agents Chemother., No. 590, Chicago, Nov. 4~6, 1981
- 25) SHIN, D. H.; J. A. FAYTER, F. BAKER, L. CAMA & B. G. CHRISTENSEN: New synthetic carbapenem antibiotics. 1,2,6-substituted-1-carbapen-2-em-3-carboxylic acid. Program and Abstracts of the 23rd Intersci. Conf. on Antimicrob. Agents Chemother., No. 333, p. 144, Las Vegas, Oct. 24~26, 1983